

## Bond Energy Pogil Answers

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bond energy 1 bond energy what makes a reaction endothermic or exothermic? 2 pogil™ activities for answers to questions 16 and 18 could be used to calculate the enthalpy change (?h) for the reaction in model 3, and then do the calculation. read this! the diagram in model 3 suggests that the production of ammonia from its elements

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bond energy is defined as the amount of energy required to break a bond. these values are positive, indicating that bond breaking is endothermic. bond energies are reported in kilojoules per mole (kJ/mol). the energy for breaking a hydrogen-hydrogen bond is 436 kJ/mol so when a hydrogen-hydrogen bond is formed the process releases 436 kJ/mol.

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chem 115 pogil worksheet - week 11 - solutions chemical bonds key questions 1. using the electronegativity values in the table above, calculate ?? for the pairs of elements forming bonds in each of the following species, and describe the type of bonding as either pure covalent, polar covalent, or ionic: h<sub>2</sub>, hcl, pbbr<sub>2</sub>, nacl, ncl<sub>3</sub>.

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